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Ali Souldozia; Ali Reza Dadrassb

^a Department of Chemistry, Islamic Azad University-Urmia Branch, Urmia, Iran ^b Chemistry Department, Faculty of Science, Urmia University, Urmia, Iran

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SILICA-GEL-CATALYZED CONVERSION OF STABILIZED PHOSPHORUS YLIDES TO BENZO[e][1,3]THIAZOCIN DERIVATIVES

Ali Souldozi¹ and Ali Reza Dadrass²

¹Department of Chemistry, Islamic Azad University–Urmia Branch, Urmia, Iran ²Chemistry Department, Faculty of Science, Urmia University, Urmia, Iran

The reaction between triphenylphosphine, dialkylacetylenedicarboxylates, and N-benzoyl thiourea derivatives leads to intermediates such as vinyltriphenylphosphonium salts, which undergo Michael addition to produce the highly functionalized phosphorus ylides containing alkyl groups in excellent yields. Using silica gel as a catalyst, the conversion of the stabilized phosphorus ylides to benzo[e][1,3]thiazocin derivatives in solvent free conditions at 90°C in fairly high yields is achieved.

Keywords Benzo[e][1,3]thiazocin derivatives; benzoyl isothiocyanate; N-benzoyl thiourea derivatives; dialkylacetylenedicarboxylates; silica gel

INTRODUCTION

Phosphorus ylides are reactive compounds that participate in many reactions of value in the synthesis of organic products. ¹⁻¹⁰ The development of the modern chemistry of natural and physiologically active compounds would have been impossible without the phosphorus ylides. ¹¹⁻¹⁶ These compounds have attained great significance as widely used reagents for molecular building blocks by the formation of carbon–carbon double bonds. ¹⁷

There are many studies on the reaction between trivalent phosphorus nucleophiles and α,β -unsaturated carbonyl compounds using phenols, imides, amides, enols, oximes, and alcohols as reaction adjuncts. Unsaturated esters such as dialkyl acetylenedicaboxylates have been employed as Michael substrates in the presence of protic reagents. ^{18–30} In this article, we report the catalytic role of silica gel powder in the conversion of stabilized phosphorus ylides to benzo[e][1,3]thiazocin derivatives in solvent-free thermal conditions in fairly high yields.

RESULTS AND DISCUSSION

Our new synthetic method that leads to the title compounds is given in Scheme 1. The reactions of *N*-benzoyl thiourea derivatives (3), which were derived from the addition of

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Address correspondence to Ali Souldozi, Department of Chemistry, Islamic Azad University-Urmia Branch, PO Box 969, Urmia, Iran. E-mail: alisouldozi@yahoo.com

$$Ph_3P$$
 + $R'OOC$ — $C\equiv C$ — $COOR'$
 $Ph_3P \oplus$
 (6)

$$\begin{array}{c} \text{R'OOC} \\ \text{Ph}_3\text{P} \circledast \\ \text{(6)} \end{array} + \begin{array}{c} \text{O} \\ \text{N} \\ \text{N} \\ \text{R} \end{array}$$

12	R N—	R'
A	diisopropylamine	CH ₃
В	diisopropylamine	CH ₃ -CH ₂
С	piperidine	CH ₃

D piperidine CH₃-CH₂

Scheme 1

secondary amines (2) to benzoyl isothiocyanate (1), with acetylenic ester (5) in the presence of triphenylphosphine (4) proceed in CH_2Cl_2 at $-5^{\circ}C$ to produce ylide (8). The TLC test indicated formation of the ylide in CH_2Cl_2 . Silica gel powder was found to catalyze the conversion of ylide (8) to benzo[e][1,3]thiazocin derivatives (12) in solvent-free conditions at $90^{\circ}C$ in fairly good yields.

The ylide (8) may result from the initial addition of triphenylphosphine (4) to the acetylenic ester (5) and concomitant protonation of the 1:1 adduct, followed by the electrophilic attack of the vinyltriphenylphosphonium cation to the anion of the benzoyl thiourea derivatives (7). The silica gel was found to catalyze the conversion of ylides (8) to vinylated compounds (9) in solvent-free conditions.

The structures were deduced from their melting points, IR, ¹H NMR and ¹³C NMR spectroscopy, and mass spectrometry.

CONCLUSION

In summary, we believe the reported method offers a mild, simple, and efficient route for the preparation of benzo[e][1,3]thiazocin derivatives. Its ease of workup, high yield, and fairly mild reaction conditions make it a useful addition to modern synthetic methodologies. Other aspects of this process are under investigation.

EXPERIMENTAL

Starting materials and solvents were obtained from Merck (Germany) and Fluka (Switzerland) and were used without further purification. TLC and NMR indicated that there is no side product. Melting points were measured on an Electrothermal 9100 apparatus and are uncorrected. IR spectra were measured on a Shimadzu IR-460 spectrometer. ¹H and ¹³C NMR spectra were measured (CDCl₃ solution) with a Bruker DRX-300 AVANCE spectrometer at 300.13 and 75.467 MHz, respectively. Elemental analyses were performed using a Heraeus CHN-O-Rapid analyzer. Mass spectra were recorded on a Finnigan-MATT 8430 mass spectrometer operating at an ionization potential of 20 eV. Flash chromatography columns were prepared from Merck silica gel powder.

General Procedure for the Preparation of Compounds 12a-d

To a solution of benzoyl isothiocyanate 1 (0.163 g, 1 mmol) and secondary amine 2 (1 mmol) in dry CH_2Cl_2 (5 mL) magnetically stirred for 2 h, triphenylphosphine 4 (0.262 g, 1 mmol) and finally a solution of acetylenic ester 5 (1 mmol) dropwise in dry CH_2Cl_2 (3 mL) was added at -5 was added C temperature over 10 min. The mixture was allowed to warm to room temperature. Silica gel powder (2 g) was added, and the solvent was evaporated. The dry materials were heated for 1 h at $90^{\circ}C$ and then placed over a column of silica gel

(10 g). The column chromatography was washed using ethyl acetate/light petroleum ether (1:10) as an eluent. The solvent was removed under reduced pressure and the products were obtained. The characterization data of the compounds (12a-d) are given below.

Selected Data for (E)-Methyl 2-((Z)-3-(Diisopropylamino)-1,6-dioxo-1H-benzo[e][1,3]thiazocin-5(6H)-ylidene)acetate (12a)

White crystals; mp: 81.4–82.9°C; Yield: 82%. IR (KBr)(v_{max} , cm^{-1}): 3020, 2965, 1706 and 1258. 1 H NMR (CDCl₃) δ_{H} : 1.42 (12H, d, 3 J_{HH} = 6.9 Hz, CH₃ of isopropyl), 3.74 (3H, s, OCH₃), 3.92–3.96 (2H, m, CH of isopropyl), 7.27 (1H, s, CH of vinylic), 7.39–7.42 and 7.81–7.84 (4H, m, arom). 13 C NMR (CDCl₃) δ_{C} : 20.02 (4C, CH₃ of isopropyl) 51.15 (1C, OCH₃), 51.44 (2C, CH of isopropyl), 127.38, 128.78 and 129.95 (5C of =CH), 128.86, 130.86 and 135.12 (4C), 160.65, 162.62 and 168.01 (3C of C=O). MS: m/z (%); 374 (M⁺, 5), 318 (60), 275(100), 261(94), 229(19), 89(18), 43(32). Anal. Calcd. For $C_{19}H_{22}N_{2}O_{4}S$ (374.45): C 60.94, H 5.92, N 7.48. Found: C 60.81, H 5.79, N 7.41%.

Selected Data for (E)-Ethyl 2-((Z)-3-(Diisopropylamino)-1,6-dioxo-1H-benzo[e][1,3]thiazocin-5(6H)-ylidene)acetate (12b)

Yellow crystals; mp: 94.0–95.1°C; Yield: 79%. IR (KBr)(v_{max} , cm^{-1}): 3031, 2966, 1700 and 1260. 1 H NMR (CDCl₃) δ_{H} : 1.26 (3H, t, $^{3}J_{HH} = 7.1$ Hz, CH₃ of Et); 1.42 (12, d, $^{3}J_{HH} = 7.0$ Hz, CH₃ of isopropyl); 3.91–3.95 (2H, m, CH of isopropyl); 4.21 (2H, q, $^{3}J_{HH} = 7.1$, CH₂ of Et), 7.27 (1H, s, CH of vinylic); 7.38–7.41 and 7.80–7.84 (4H, m, arom). 13 C NMR (CDCl₃) δ_{C} : 14.34 (1C, CH₃ of Et); 20.04 (4C, CH₃ of isopropyl); 51.14 (1C, CH₂ of Et); 60.31 (2C, ch of isopropyl); 127.32, 128.68 and 129.99 (5C of =CH); 128.79, 130.90 and 135.28 (4C), 159.76, 162.25 and 167.98 (3C of C=O). MS: m/z (%); 388 (M⁺, 3), 332 (56), 289 (100), 244 (81), 160 (15), 101 (21), 77(25), 58 (31), 45 (28). Anal. Calcd. For C₂₀H₂₄N₂O₄S (388.48): C 61.83, H 6.23, N 7.21. Found: C 61.75, H 6.15, N 7.19%.

Selected Data for (E)-Methyl 2-((Z)-1,6-Dioxo-3-(piperidin-1-yl)-1H-benzo[e][1,3]thiazocin-5(6H)-ylidene)acetate (12c)

White crystals; mp: 99.5–100.6°C; Yield: 75%. IR (KBr)(v_{max} , cm^{-1}): 3015, 2997, 2962, 1715 and 1245. ¹H NMR (CDCl₃) $\delta_{\rm H}$: 1.68–1.72 (6H, m, CH₂ of piperidine); 3.57–3.59 (4H, m, CH₂ of piperidine); 3.74 (3H, s, OCH₃); 7.27 (1H, s, CH of vinylic); 7.39–7.41 and 7.75 (4H, m, arom). ¹³C NMR (CDCl₃) $\delta_{\rm C}$: 22.97, 23.76 and 49.17 (5C, CH₂); 51.53 (1C, OCH₃); 127.54, 128.90 and 129.76 (5C of =CH); 128.79, 130.86 and 134.90 (4C); 160.38, 162.44 and 170.90 (3C of C=O). MS: m/z (%); 358 (M⁺, 3), 300 (100), 275(77), 167(18), 149 (40), 85 (21), 71 (19), 58 (47), 43 (95). Anal. Calcd. For C₁₈H₁₈N₂O₄S (358.41): C 60.32, H 5.06, N 7.82. Found: C 60.27, H 4.98, N 7.75%.

Selected Data for (E)-Ethyl 2-((Z)-1,6-Dioxo-3-(piperidin-1-yl)-1H-benzo[e][1,3]thiazocin-5(6H)-ylidene)acetate (12d)

Viscous oil; Yield: 71%. IR (KBr)(v_{max} , cm^{-1}): 3027, 2936, 1708 and 1243. ¹H NMR (CDCl₃) $\delta_{\rm H}$: 1.24 (3H, t, ${}^3J_{\rm HH}$ = 7.1 Hz, CH₃ of Et); 1.68–1.70 (6H, m, CH₂ piperidine); 3.57–3.58 (4H,m, CH₂ of piperidine); 4.20 (2H, q, ${}^3J_{\rm HH}$ = 7.1 Hz, CH₂ of Et); 7.27 (1H, s, CH vinylic), 7.37–7.39 and 7.73–7.76 (4H, m, arom). ¹³C NMR (CDCl₃) $\delta_{\rm C}$: 14.25 (1C, CH₃); 23.76, 23.99 and 49.17 (5C, CH₂); 60.46 (1C, OCH₂), 127.49, 128.82 and 129.79

(5C, =CH); 127.83, 130.86 and 135.01 (4C); 159.99, 162.05 and 170.84 (3C of C=O). MS: m/z (%); 372 (M⁺, 4), 314 (100), 289 (71), 269 (20), 242 (10), 217 (8), 133 (38), 85 (18), 45 (21). Anal. Calcd. For $C_{19}H_{20}N_2O_4S$ (372.44): C 61.27, H 5.41, N 7.52. Found: C 61.24, H 5.37, N 7.48%.

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